

# Improvement of Traditional HPLC Methods by Combining Temperature with a New Generation HPLC columns

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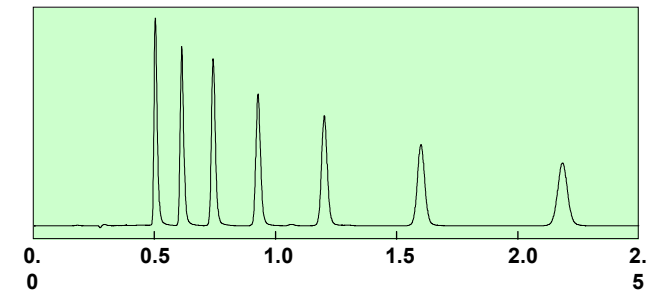
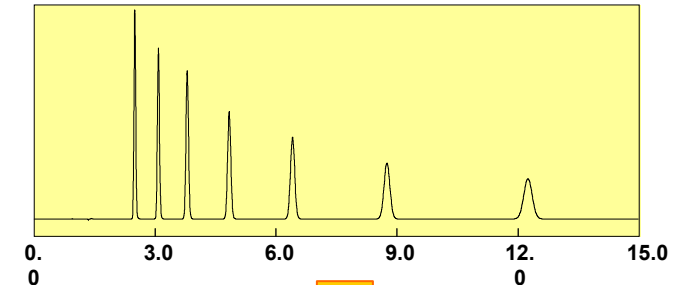


# Goal: Enhancing Productivity

## Faster - More speed

→ Shorter run time while maintaining resolution

- Higher linear velocity
- Shorter column length



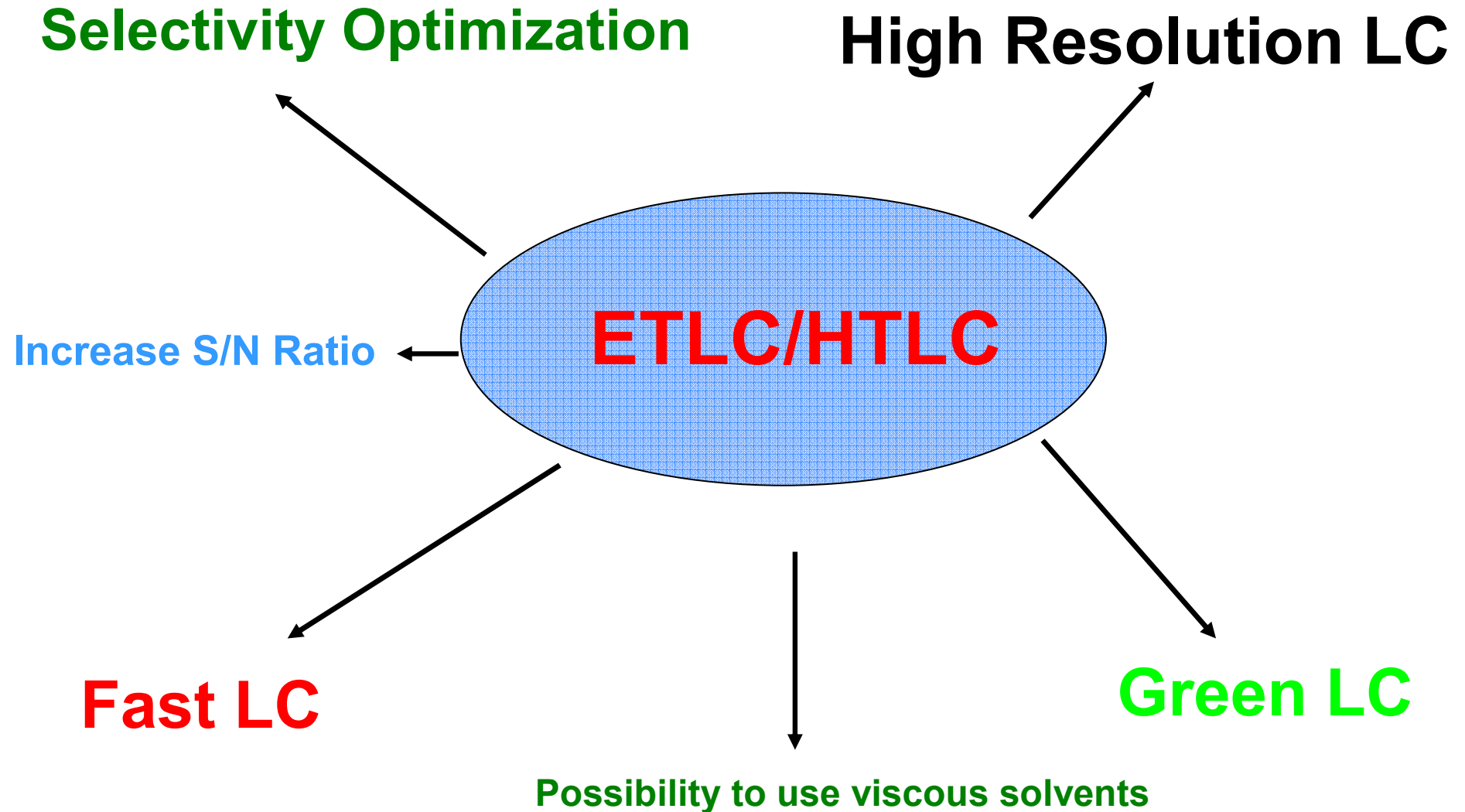
## Better - More resolution

→ Better resolution while maintaining run time

- Higher theoretical plate numbers
- Different selectivity



# Temperature Driven LC





# Why Fast LC?

$$\Delta G^{\circ} = \Delta H^{\circ} - T\Delta S^{\circ}$$

$$\Delta G^{\circ} = -RT \ln\left(\frac{k}{\phi}\right)$$

$$\ln k = -\frac{\Delta H^{\circ}}{RT} + \frac{\Delta S^{\circ}}{R} \ln \phi$$

K: Retention Factor

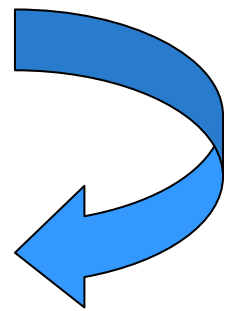
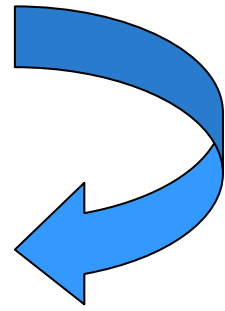
$\Delta H^{\circ}$ : Enthalpy

$\Delta S^{\circ}$ : Entropy

R: Gas Constant

T: Temperature

$\phi$ : Phase Ratio





# Why Fast LC?

Hydrophobic retention is in most cases exothermic, therefore both variables are negative

$$\text{Ln}k = -\frac{\Delta H^{\circ}}{RT} + \frac{\Delta S^{\circ}}{R} \text{Ln}\phi$$

K: Retention Factor

$\Delta H^{\circ}$ : Enthalpy

$\Delta S^{\circ}$ : Entropy

R: Gas Constant

T: Temperature

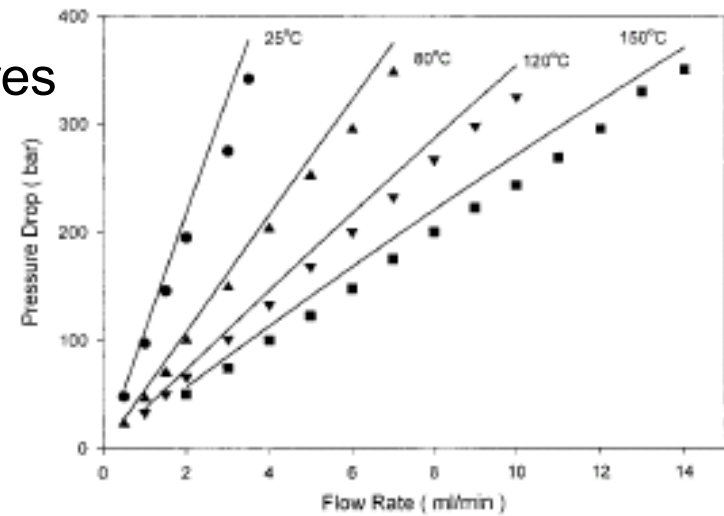
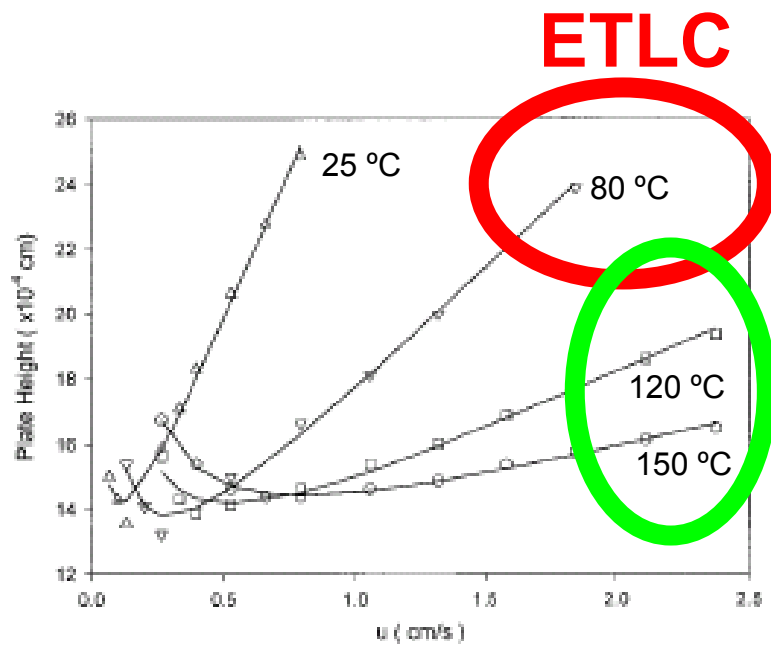
$\Phi$ : Phase Ratio

**T ↑ = K ↓**



# Favorable Properties with Temperature Driven LC

Reduced backpressure at elevated Temperatures  
(Increase your flow!!)

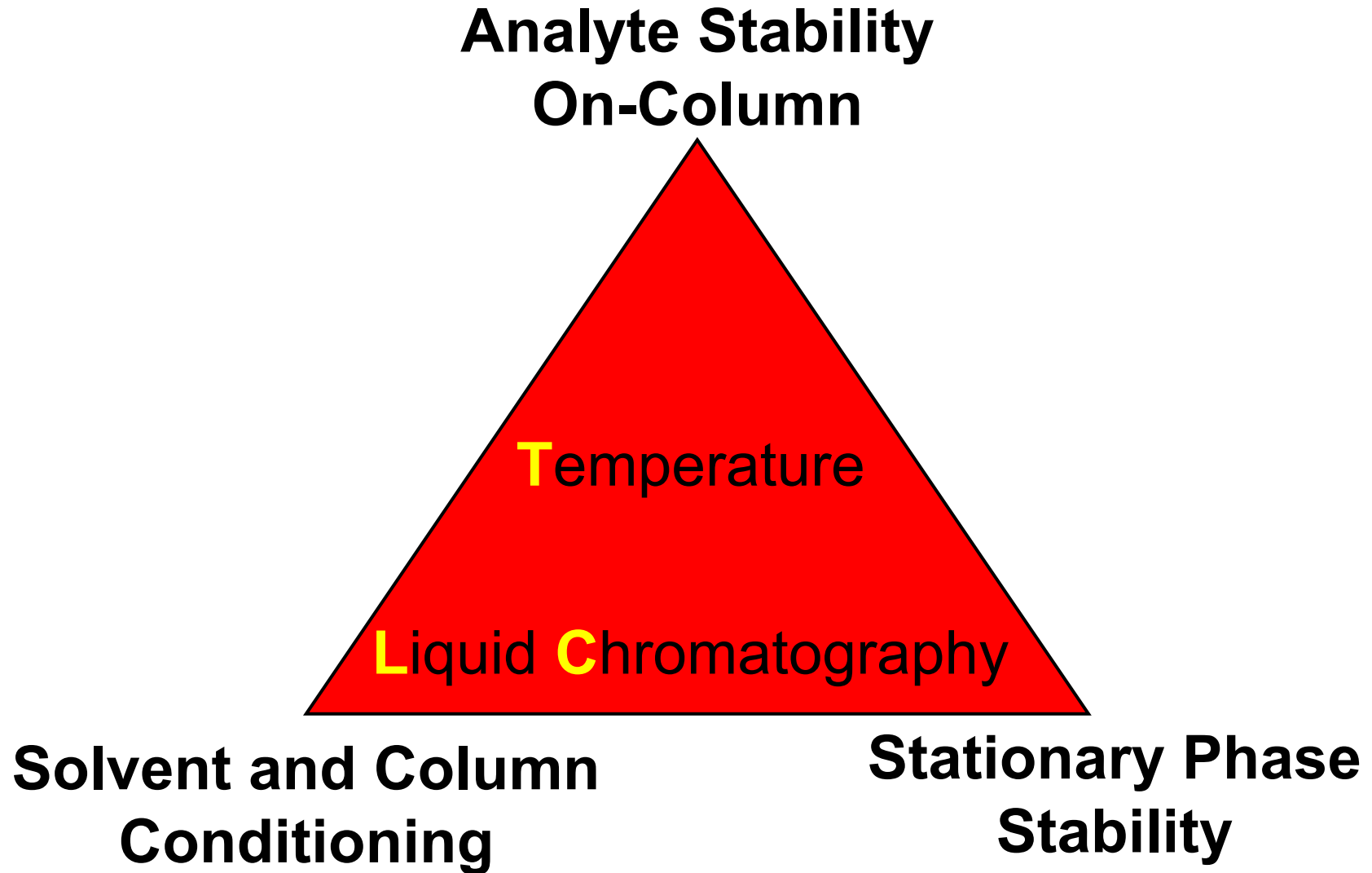


**HTLC**

Mass transfer increases as temperature increases  
(Increase your flow!!)



# Requirements for Temperature Driven LC





# Column Technology Is There A Solution?

A Wide Range Of Particle And Pore  
Sizes  
High Chromatographic Efficiency  
High Bed Stability

**Silica Particles**

Low Chemical Stability  
Low Thermal Stability  
Short Lifetimes



Flexibility In Chemical Design  
High Chemical Stability  
No Silanol Tailing

**Polymeric Particles**

Low Mechanical Stability  
Low Mass Transfer Properties  
Low Chromatographic Performance





# Is There A Solution?

A Wide Range Of Particle And Pore Sizes  
High Chromatographic Efficiency  
High Bed Stability

Silica Particles

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Polymeric Particles

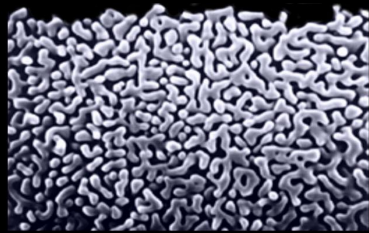
Low Mechanical Stability  
Low Mass Transfer Properties  
Low Chromatographic Performance



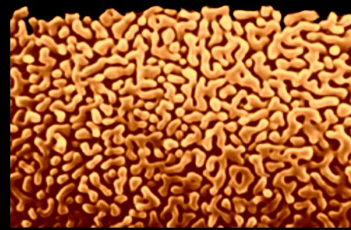
Shant  
LABORATORIES



# Silica-Polymer Hybrid Technologies



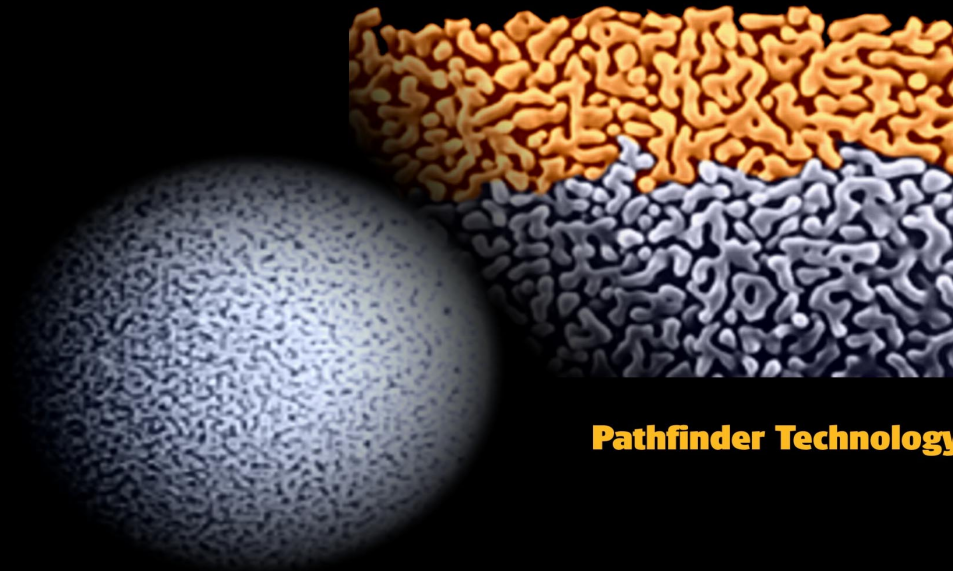
**Silica Technology**



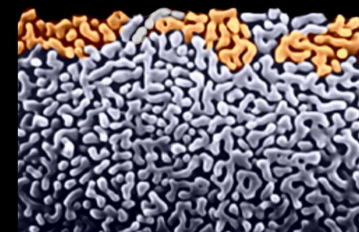
**Polymer Technology**



**Hybrid Technology**



**Pathfinder Technology**



**Polymer Grafted Technology**



**Polymer**



**Silica**



# Pathfinder Family

**Pathfinder  
r  
Standard<sup>®</sup>**



Follow up the rules of  
traditional LC

**Mixed Mode  
Columns**

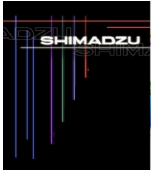
**MR** **EP**

**PS**

**AS** **AP**

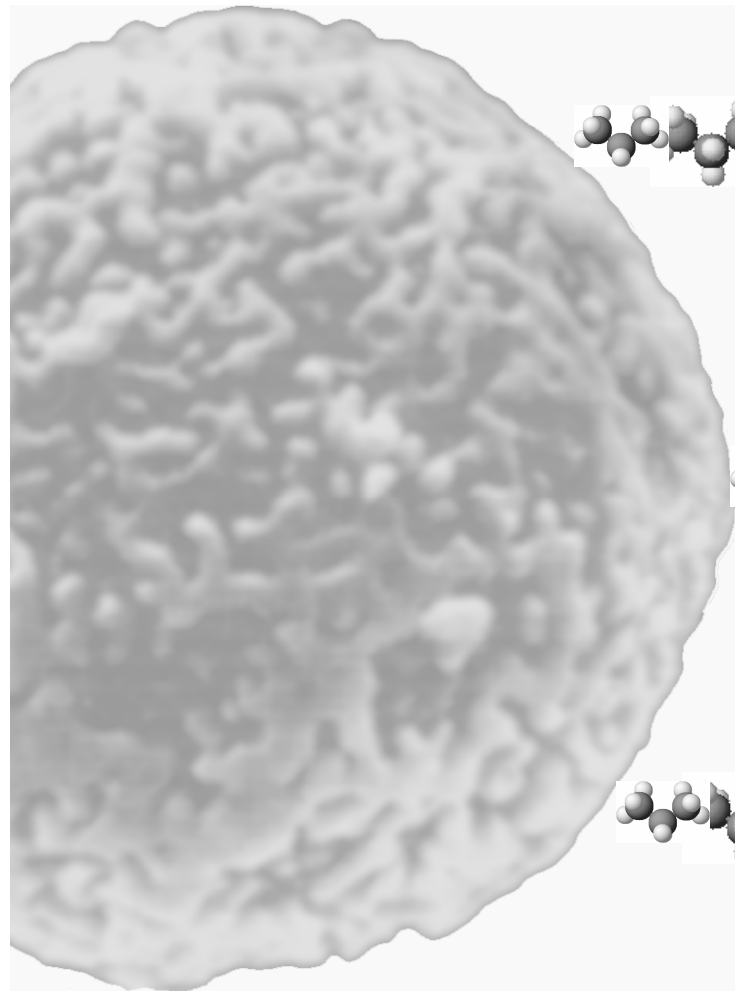


For difficult separations  
as alternative to  
traditional RP columns



# Pathfinder Standard

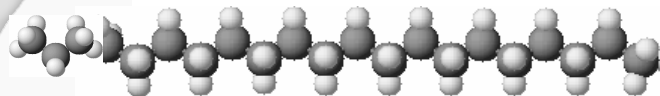
## Surface Chemistry



C18 chain



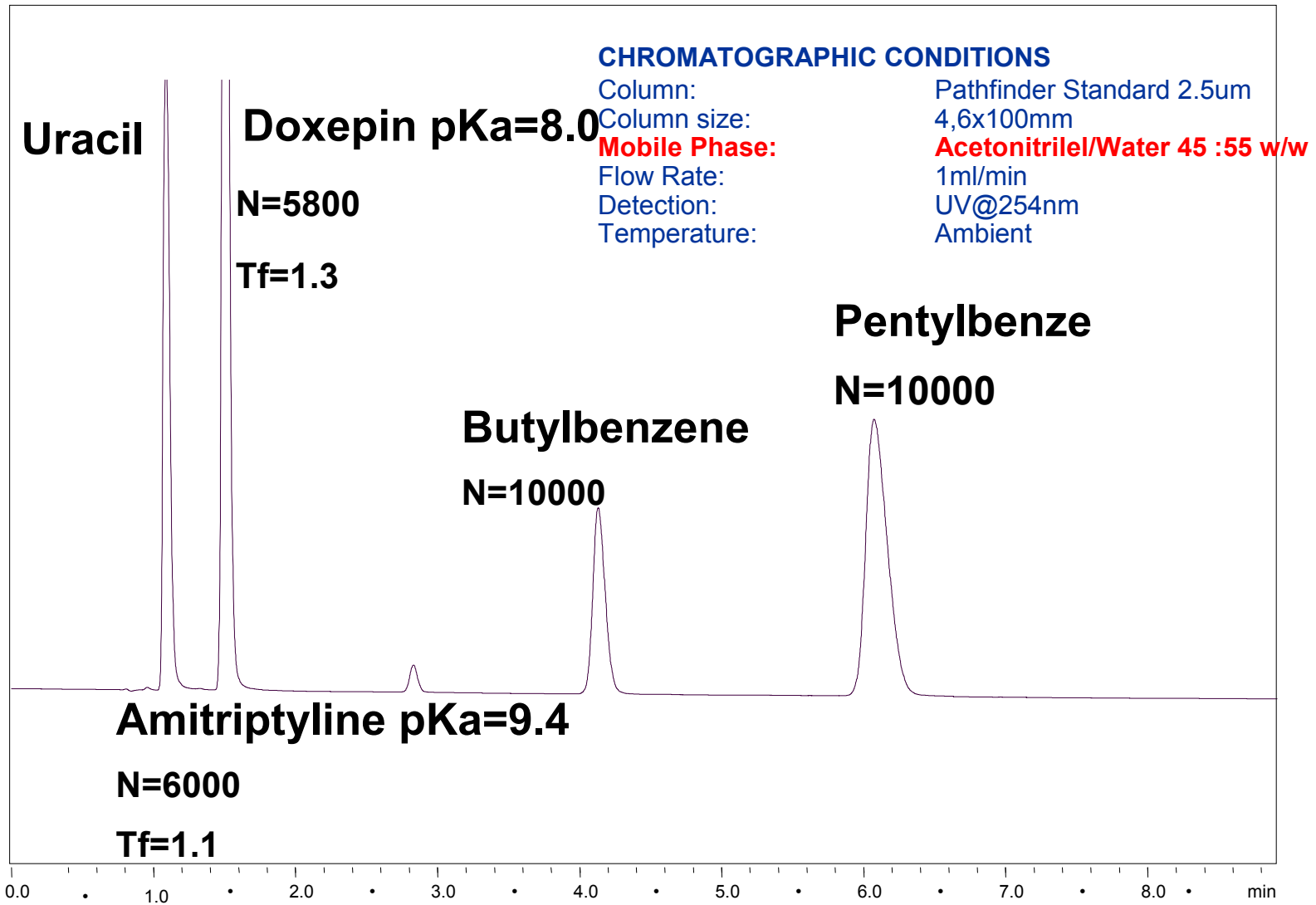
Hydrophobic  
Interaction



Compounds analysed largely  
by dispersive interaction of  
analyte and stationary phase



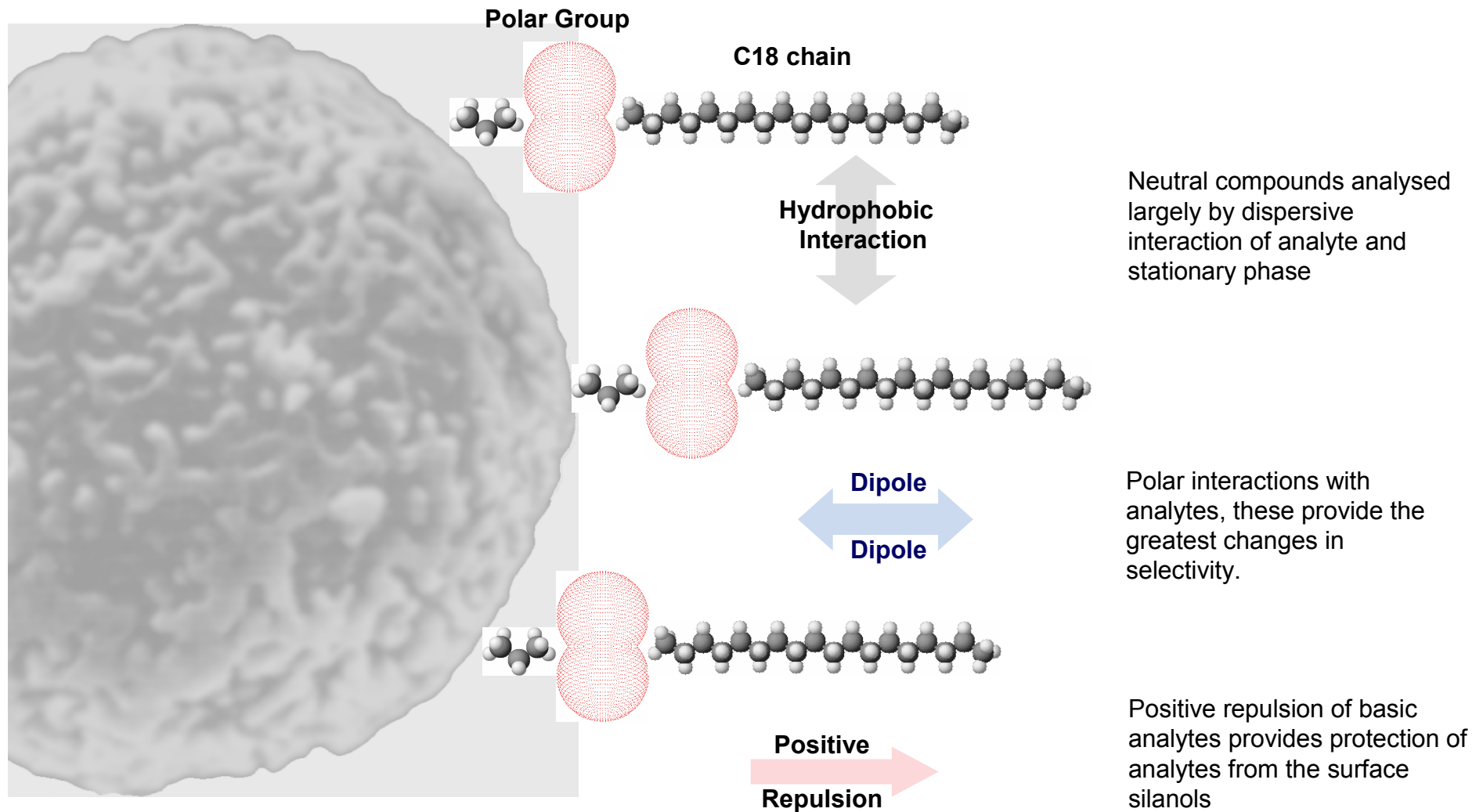
# Pathfinder Standard No Buffer Needed For Basic Molecules!!!

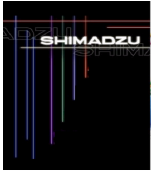




# Pathfinder Mixed Mode

## Surface Chemistry

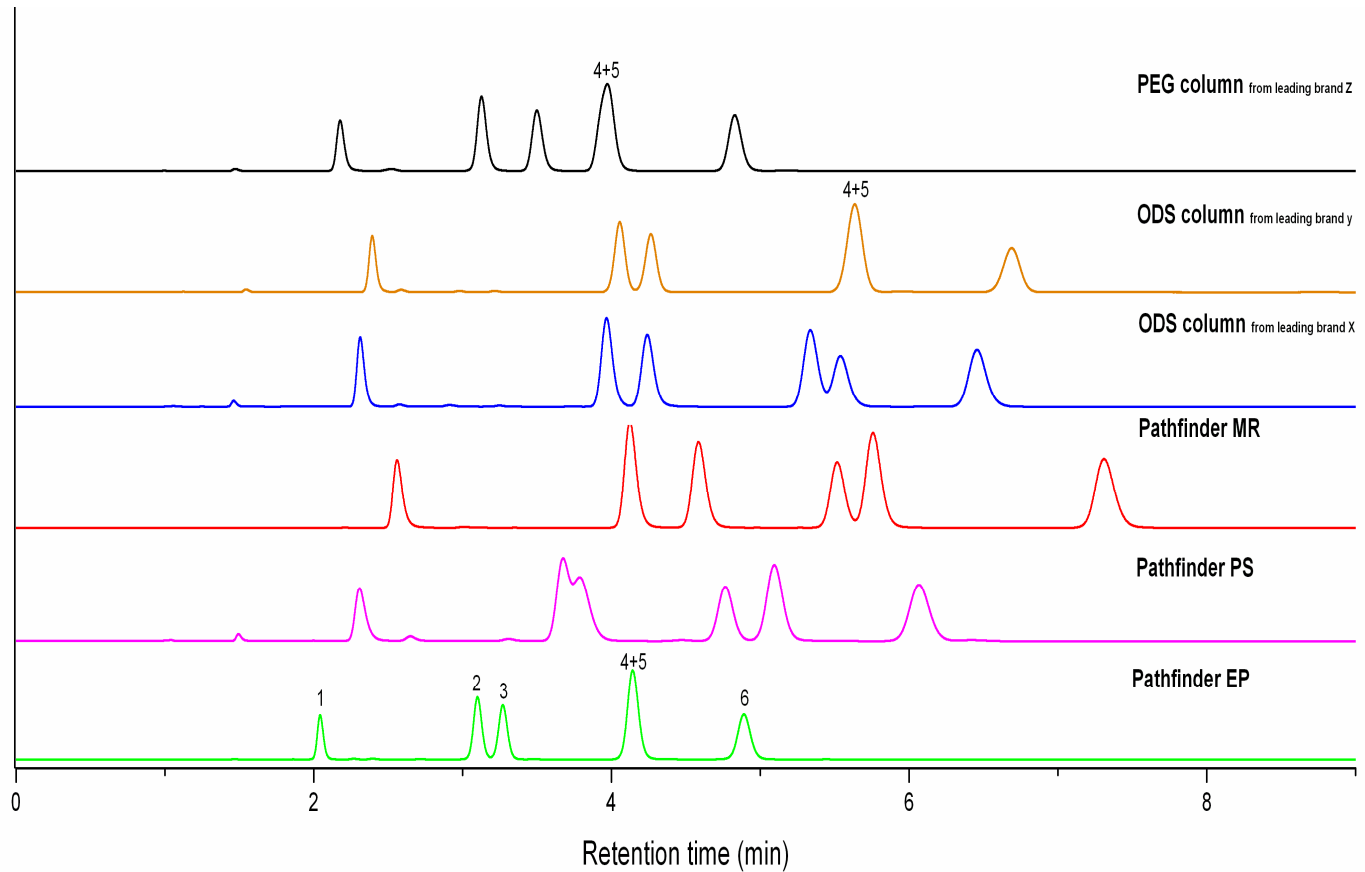




# Application: Separation of Steroids with Pathfinder Mixed Mode

## Compounds:

1. Hydrocortisone
2. 11- $\alpha$ -hydroxyprogesterone
3. Cortisone Acetate
4. Dioxycorticosterone
5. 11-ketoprogesterone
6. 17- $\alpha$ -hydroxyprogesterone

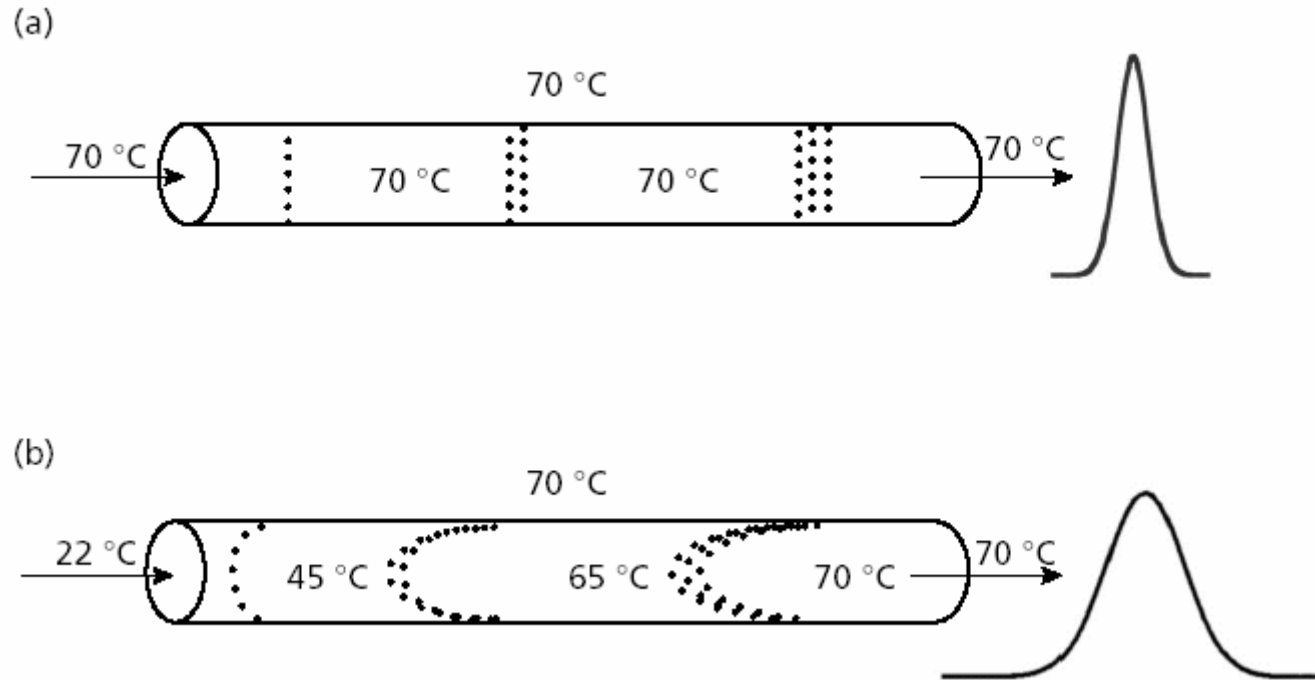


## CHROMATOGRAPHIC CONDITIONS

Column size: 4,6x150mm 5 $\mu$ m  
Mobile Phase: Acetonitrile/Water 50 :50 w/w  
Flow Rate: 1ml/min  
Detection: UV@254nm  
Temperature: Ambient



# Mismatch due to Temperature Gradients

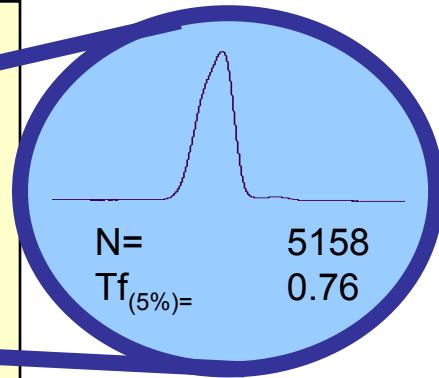
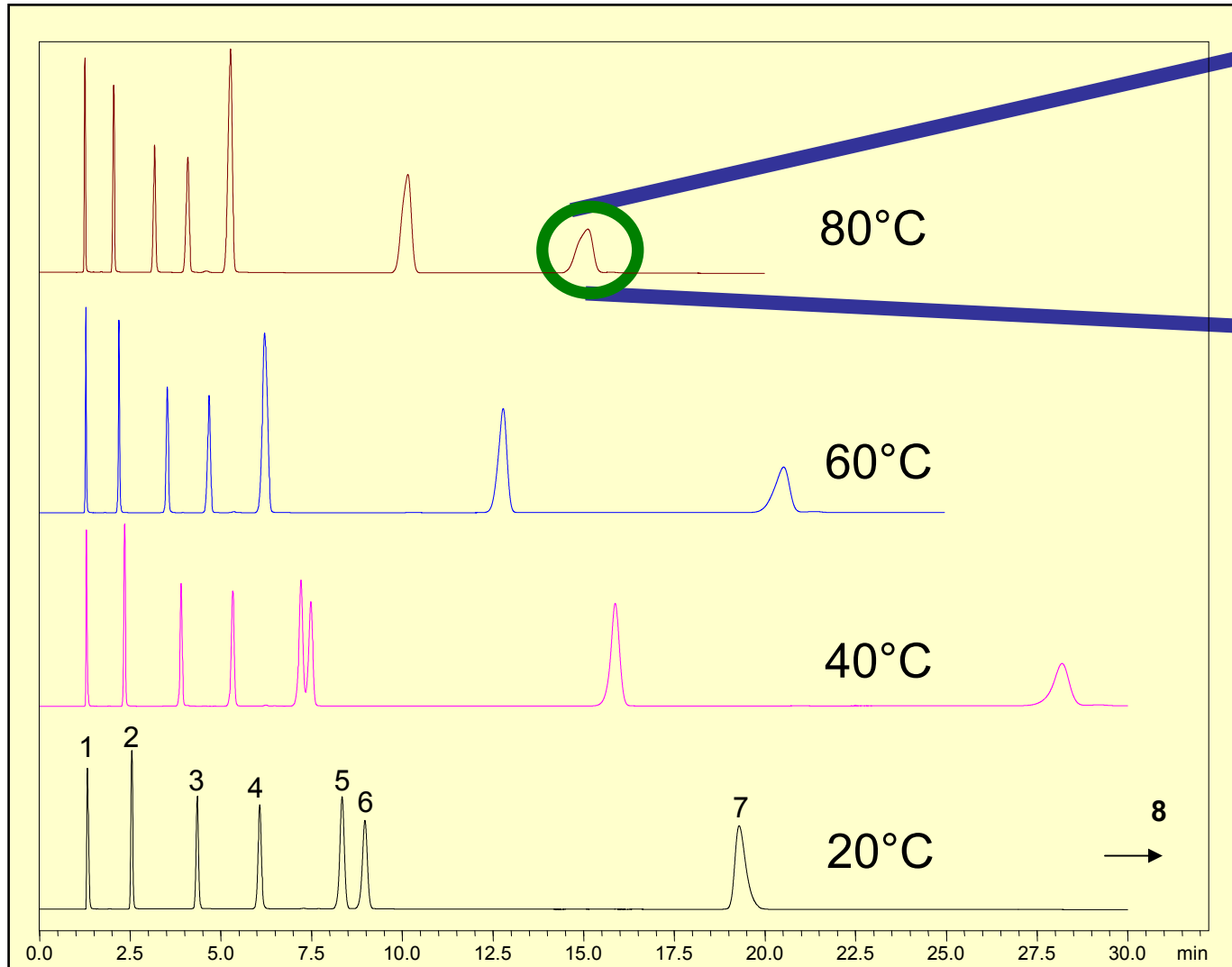


Peak shape decreases drastically when temperature gradients occurs inside the column.





# Temperature Effects No Solvent Pre Heating in **ETLC**



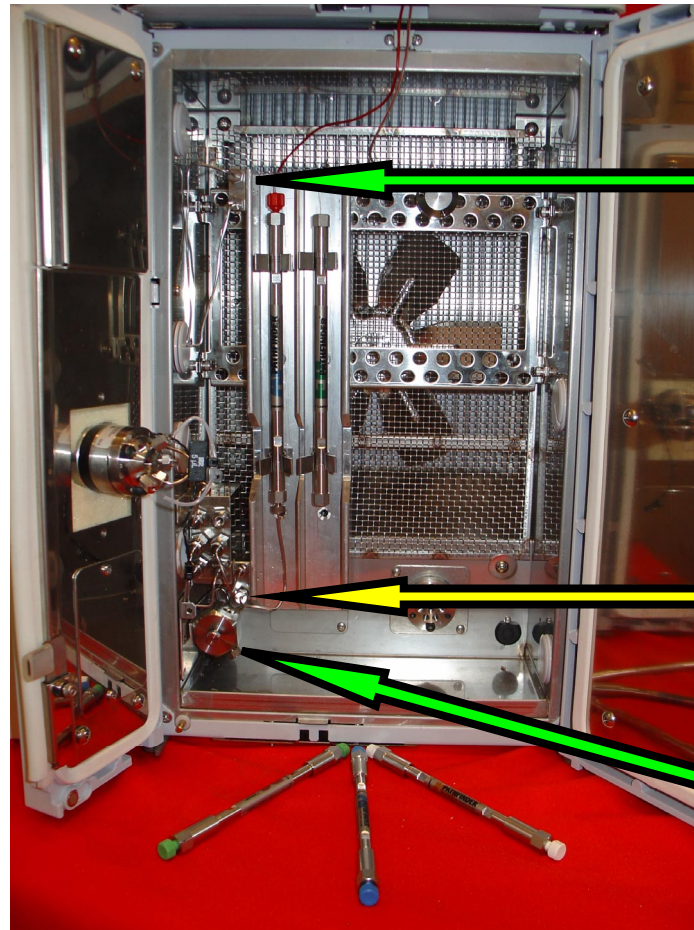
1. Uracil
2. Acetophenone
3. Benzene
4. Toluene
5. Ethylbenzene
6. Naphtalene
7. Butylbenzene
8. Pyrene



# Heat Block Assembly CTO10ASvp Integrated in CTO20AC Column Oven

## Points of attention:

- 0.13 mm i.d. SUS tubing
- Direct column connection with flow cell
- Ch.1/Ch.2 each 10 $\mu$ l d dead volume
- Assembly P/N 228-35158-91



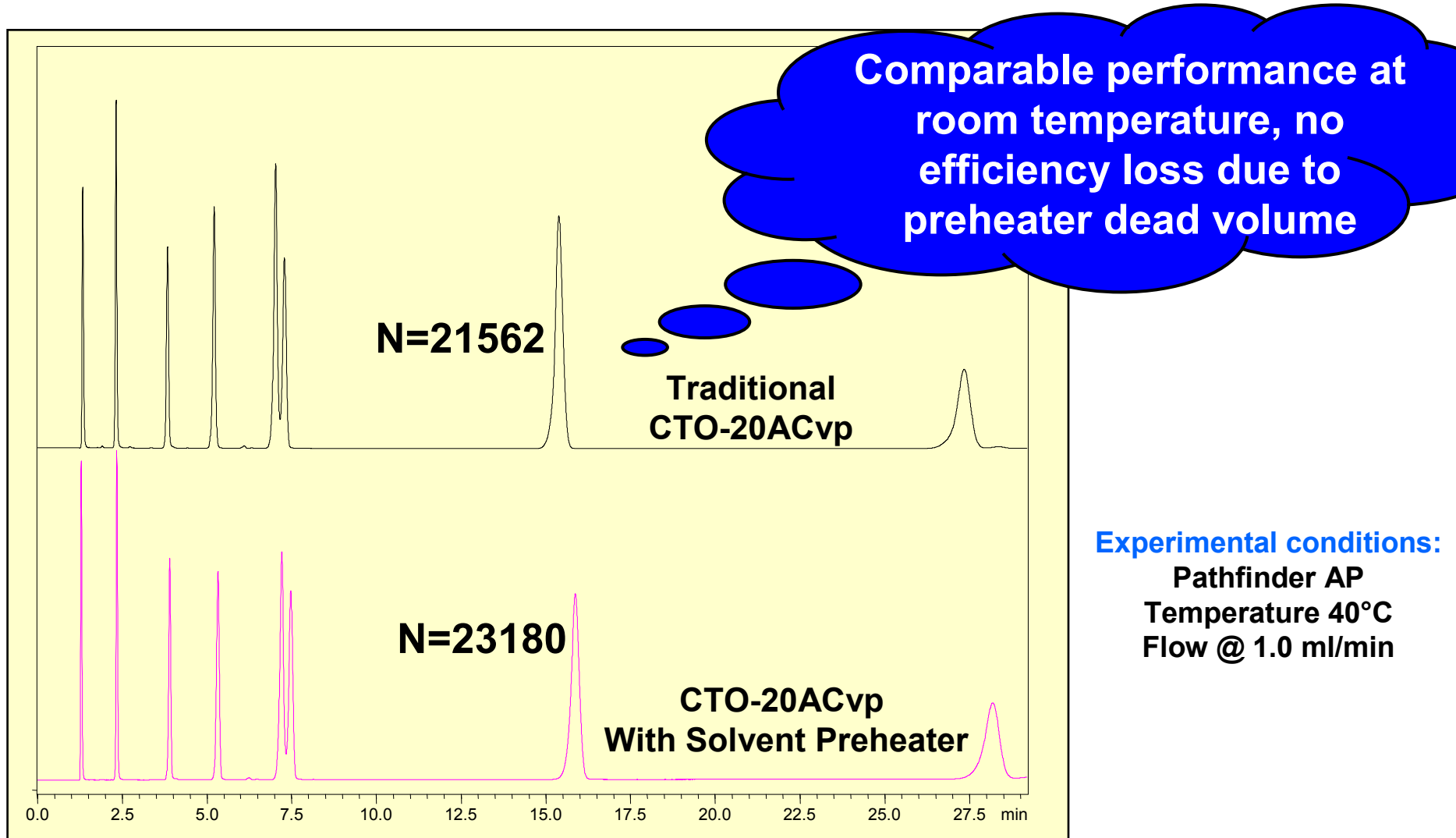
Ch 1. Solvent preheated prior to injection

Ch 2. Solvent/sample preheating before entering the column

Solvent preheating inside mixing chamber



# Performance Comparison CTO-20ACvp with and without Solvent Pre Heater





## Pre Heater Capacity @ 85°C ~Plate number vs flow and temperature~

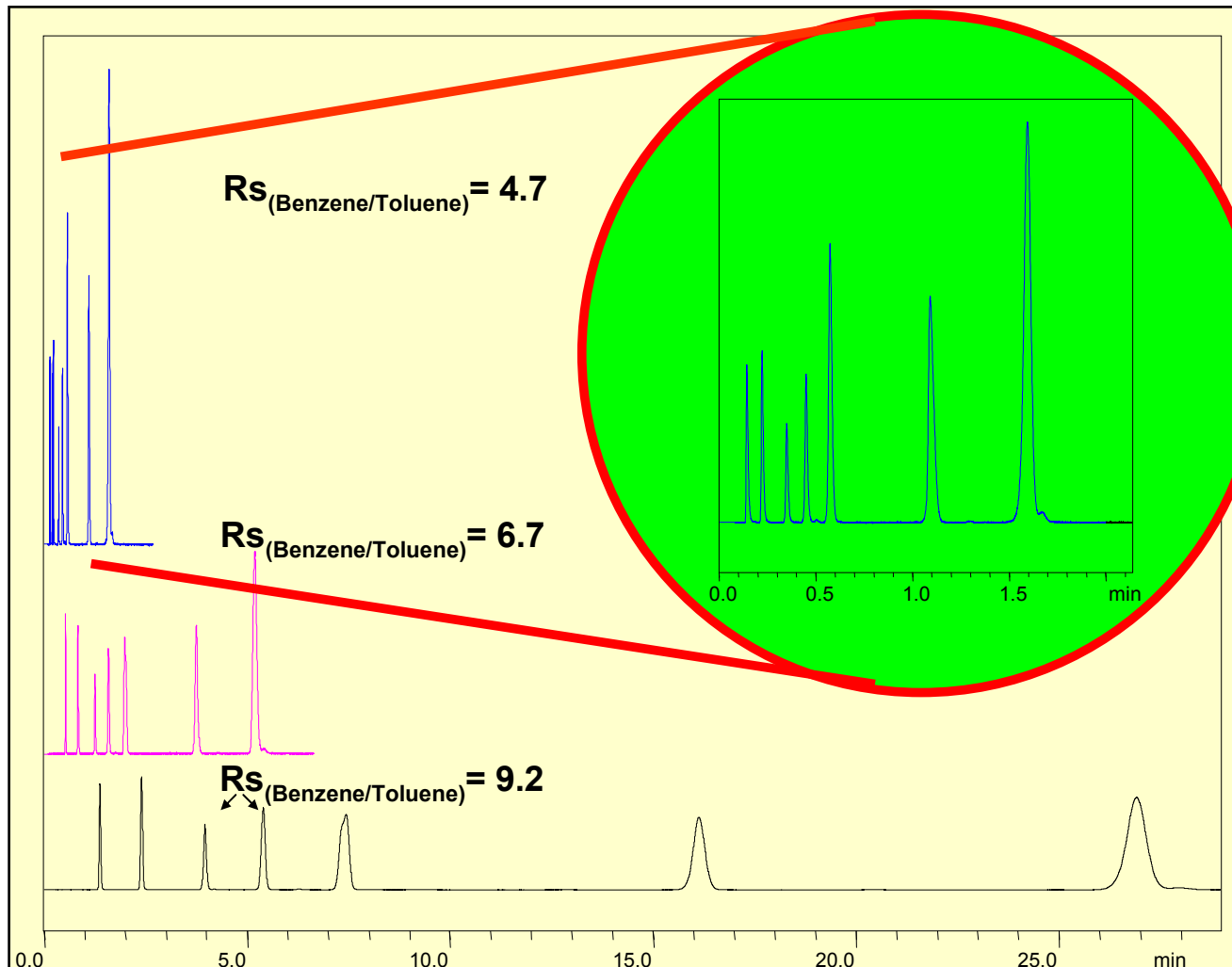
<b>Name</b>	<b>40°C 1.0 ml/min</b>	<b>85°C 1.0 ml/min</b>	<b>85°C 1.5 ml/min</b>	<b>85°C 2.0 ml/min</b>
<b>Uracil</b>	4672	7292	5972	5370
<b>Acetophenone</b>	12785	13465	12367	11299
<b>Benzene</b>	19417	17893	18479	17873
<b>Toluene</b>	21581	20183	20622	19636
<b>Naphthalene</b>	NA	NA	NA	NA
<b>Ethylbenzene</b>	NA	NA	NA	NA
<b>Butylbenzene</b>	23180	21796	23912	25246
<b>Pyrene</b>	23363	24187	25621	25909

**No loss in efficiency over the entire flow  
and temperature range!!**



# Speed up your Analysis

## 18 Times Faster!!



**Pathfinder AS**  
**2.5  $\mu\text{m}$  x 100Å**  
4.6 mm ID x 50 mm  
Flow @ 3.0 ml/min  
Temperature 85°C  
Pressure 170 bar

**Pathfinder AS**  
**3,5  $\mu\text{m}$  x 100Å**  
4.6 mm ID x 150 mm  
Flow @ 2.5 ml/min  
Temperature 85°C  
Pressure 120 bar

**Pathfinder AS**  
**5.0  $\mu\text{m}$  x 100Å**  
4.6 mm ID x 150 mm  
Flow @ 1.0 ml/min  
Temperature 40°C  
Pressure 80 bar



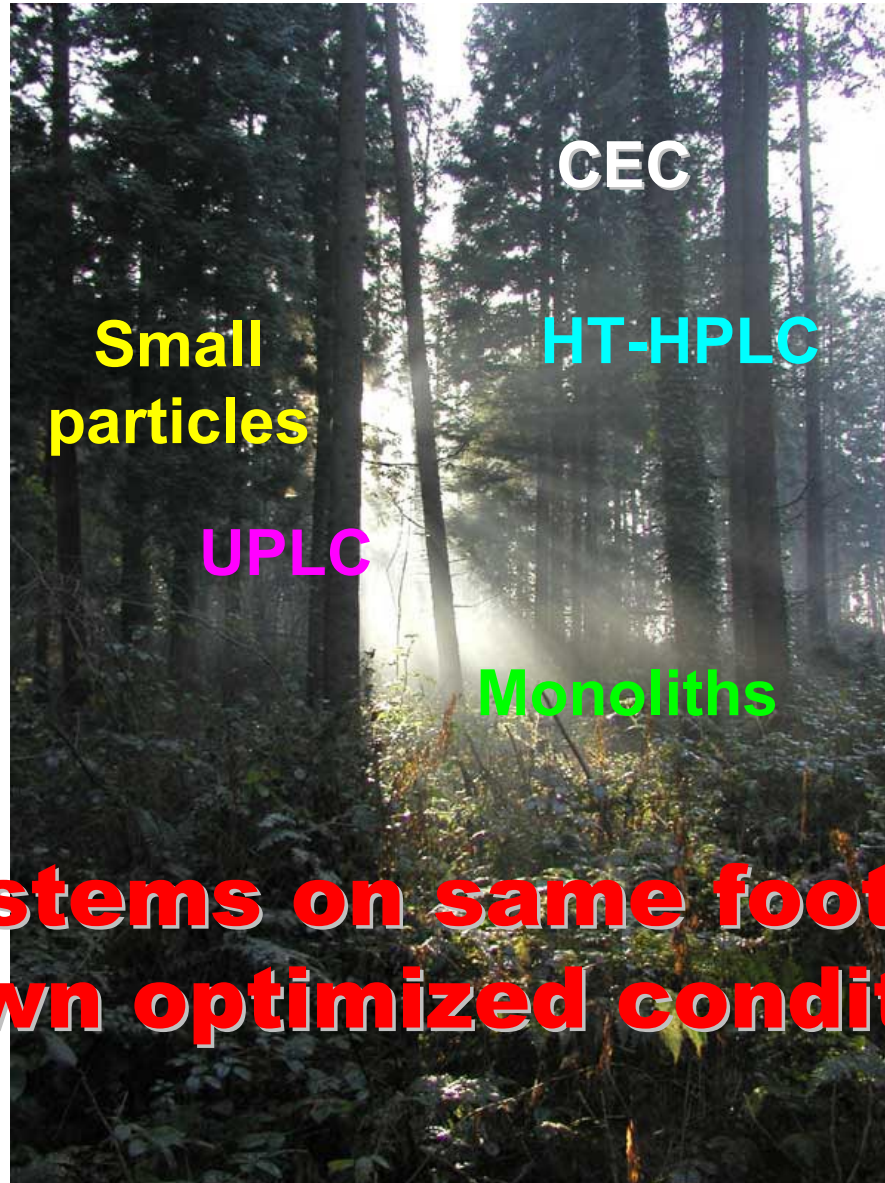
# What About:

## **High Temperature Liquid Chromatography**





# How keeping the overview ?

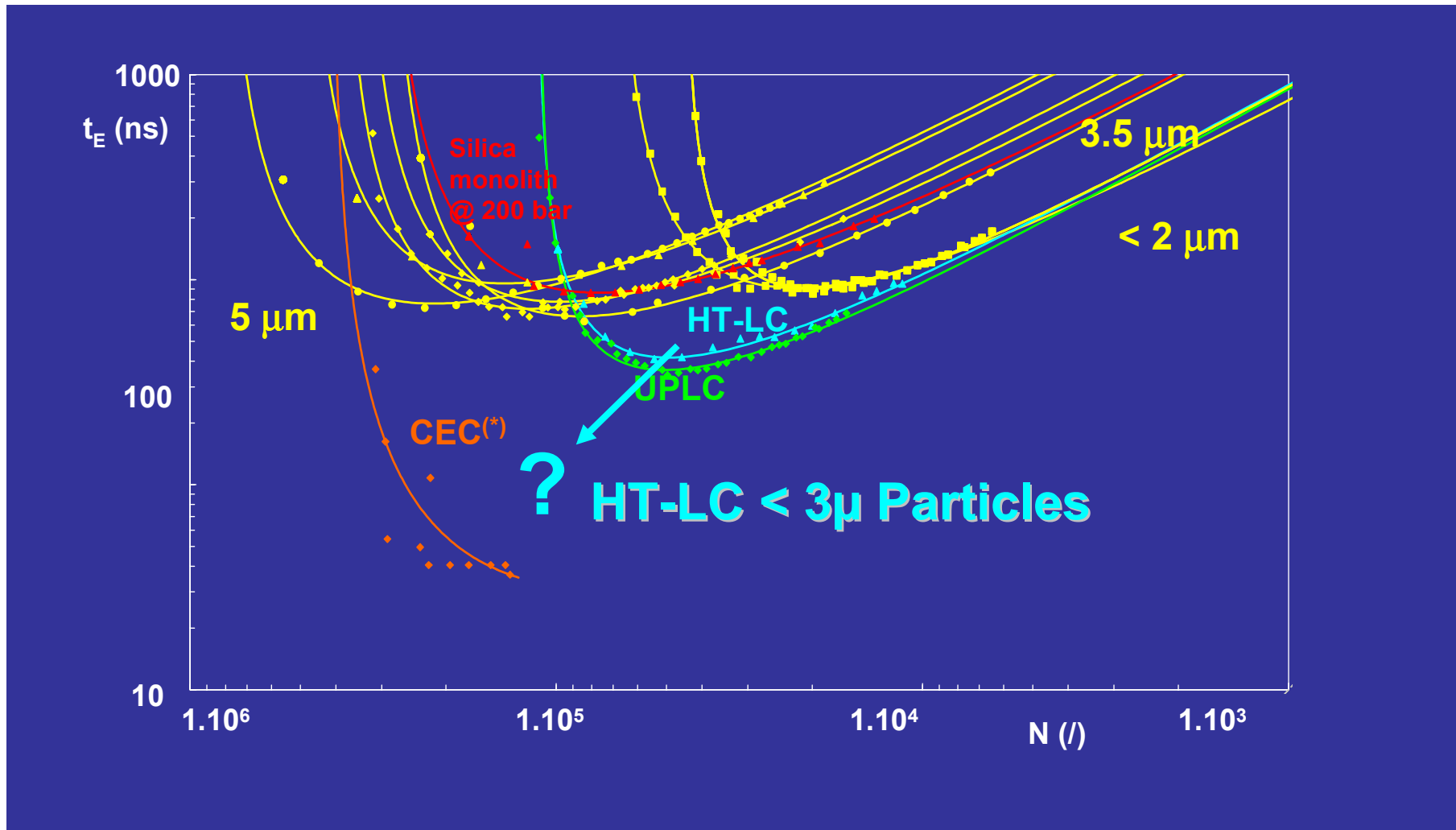


**Compare systems on same foot, each for their own optimized conditions**



# Monoliths ↔ UPLC ↔ HT-HPLC ↔ CEC

Plots Provided by Prof. Desmet (VUB)



(\*)R. Stol, W. Th. Kok and H. Poppe, J. Chrom A, 853 (1999) 45-54  
Fluorene on 7  $\mu\text{m}$  wide porous RP-HPLC beads (400 nm)





# Application HTLC: Preservatives

## CHROMATOGRAPHIC CONDITIONS

Column: Pathfinder MR 2.5 $\mu$ m

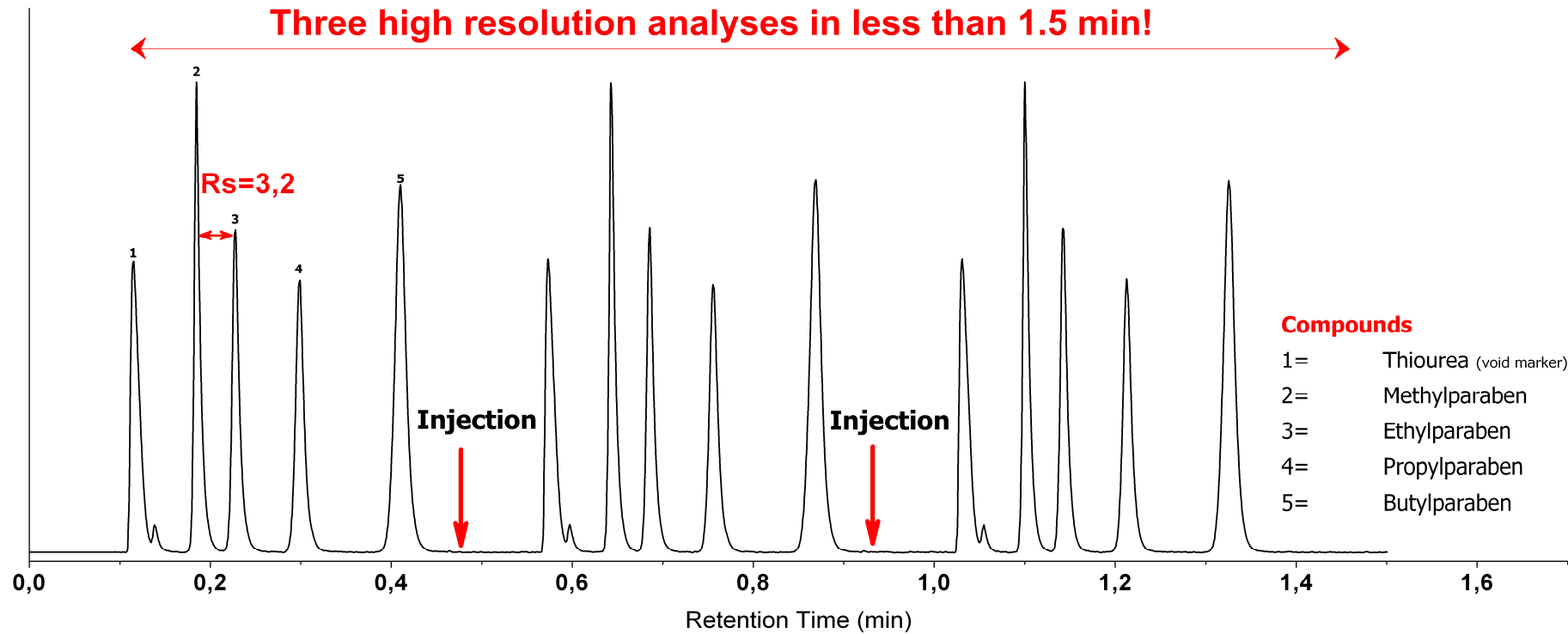
Column size: 4,6x50mm

Mobile Phase: Acetonitrile/Water 45 :55 w/w

Flow Rate: 5ml/min

Detection: UV@254nm

**Temperature: 110 degrees**





# Shimadzu's Temperature Approach

## PATHFINDER<sup>®</sup>

### ETLC with **Shimadzu Prominence**

- Fast LC analysis
- Fast LC gradients
- No loss in efficiency below 85°C using 4.6 mm ID columns
- Appropriate heating capacity up to 3.0 ml/min
- Temperature & pH range compatible with Pathfinder column technology
- No need for special equipment, cost saving!

Near future:

**High Temperature Separations**

Pathfinder<sup>®</sup> is a registered trademark of Shant Laboratories sa/nv



## Acknowledgements

### **Shimadzu Benelux**

Ria Emonds

Johan Scholtens

Vincent Goudriaan

### **Shimadzu USA**

Dr. Masayuki Nishimura

### **Shant Laboratories**

Dr. George Abrahamyan

### **Shimadzu Europe**

Dr. Björn Erxleben

Robbert



**For information and applications**

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